# STEREOSPECIFIC PREPARATION OF TERMINAL VINYLLITHIUM DERIVATIVES BY

## Br/Li-EXCHANGE WITH T-BUTYLLITHIUM

Helmut Neumann and Dieter Seebach

### Institut für Organische Chemie der Justus Liebig-Universität, Heinrich-Buff-Ring 58 D-6300 Gießen

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Hitherto, there is no generally applicable procedure of preparing the vinyl lithium compounds  $\frac{2}{2}$  from terminal bromides  $\frac{1}{2}$  by Br/Li-exchange with RLi<sup>1, 2)</sup>. Previous attempts caused dehydrobromination with formation of products derived from acetylenes<sup>3)</sup>. Reagents  $\frac{2}{2}$  are therefore made by reacting Li metal with the corresponding bromides<sup>4)</sup>, by carrying out R<sub>3</sub>Sn/Li transmetalations<sup>3, 5)</sup> or by switching to the less readily available vinyliodides<sup>1, 6)</sup>. Also, vinylic organomagnesium<sup>7)</sup> and -aluminium<sup>8)</sup> derivatives are used for this type of stereospecific olefin synthesis.

We have now found, that terminal alkenyllithiums  $\frac{2}{2}$  can be generated from the easily accessible<sup>9</sup> bromides  $\frac{1}{2}$  in high yields and with clean retention of configuration by employing 2 equivalents of <u>t</u>-butyllithium<sup>10</sup> at -120°C in the Trapp-solvent mixture<sup>11</sup>.

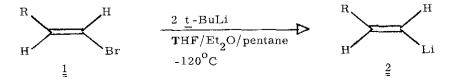
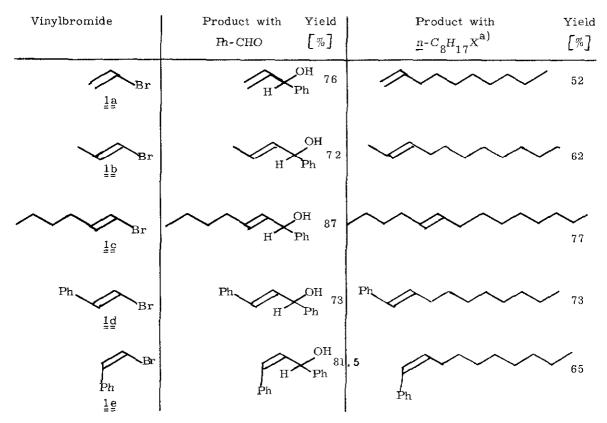


Table 1 shows the products with yields obtained from different vinyl bromides by trapping  $\frac{2}{2}$  with benzaldehyde and 1-halooctanes. In the case of the cis-trans isomeric  $\omega$ --bromo-styrenes  $\frac{1}{2}$  and  $\frac{1}{2}$  we have proved by GC-analysis that the E-olefin synthesized from  $\frac{1}{2}$  and bromooctane did not contain any Z-isomer and that  $\frac{1}{2}$  was converted to better than 99.7% configurationally pure Z-olefin<sup>12)</sup>.

The unsubstituted vinyllithium  $(\frac{2}{2}\frac{1}{2})$  and E-1-lithio-1-hexene  $(\frac{2}{2}\frac{1}{2})$  were also derivatized with the heteroelectrophiles diphenyl and dimethyl disulfide, iodine, and chloro

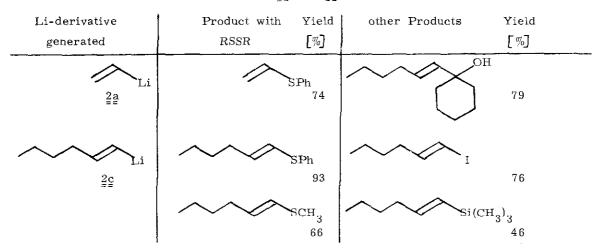
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Table 1: Stereospecific coupling of terminal vinylbromides  $\frac{1}{2}$  with benzaldehyde and 1-halooctane through the vinyllithiumderivatives  $\frac{2}{2}$ .



a) With  $\lim_{n \to \infty} and \lim_{n \to \infty} X=I$ , in all other cases X=Br

Table 2: Miscellaneous conversions of la and lc



trimethylsilane; cyclohexanone and  $2c_{zz}$  gave the expected allylic alcohol. The yields and product structures are shown in Table 2.

Since there is no organic halide present after the Br/Li-exchange and the mutual destruction of <u>t</u>-butyl bromide and <u>t</u>-butyllithium at  $-120^{\circ}$ C, the solutions of <u>2</u> can be warmed up to room temperature if desired for reactions with electrophiles. To avoid acetylene formation, however the bath temperature during the exchange step  $\frac{1}{2} \rightarrow \frac{2}{2}$ , must not be above  $-110^{\circ}$ C.

All yields given in the tables are those of distilled products; the IR and NMR data are fully compatible with the structures shown; new compounds analyzed correctly, the physical properties of known products agreed with the literature data.

<u>Procedure</u>: A magnetically stirred solution of 10 mmoles of  $\frac{1}{2}$  in 42 ml of Trapp-mixture (THF/Et<sub>2</sub>O/pentane 4:1:1) was cooled under an argon atmosphere in a -120°C bath [ligroin (30-50), isopropanol, acetone (4:1:1)/liquid N<sub>2</sub>] and combined within 10 min . with 20 mmoles of <u>t</u>-butyllithium (in pentane). The temperature was kept between -120°C and -110°C for 1 hr, and then raised to -90°C. 9.9 mmoles of benzaldehyde were added, and stirring was continued for 15 min at -78°C and for 20 min at room temperature. The mixture was quenched by pouring into a separatory funnel containing 10 mmoles of acetic acid, saturated aqueous NaCl solution and methylene chloride<sup>10</sup>.

To achieve alkylation, sequential addition of 9 mmoles of 1-bromo or 1-iodo octane and of 50 ml of THF at  $-90^{\circ}$ C was followed by stirring at room temperature for 18 hr.

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